Acid - Base What You Need To Know

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Introduction

- What tests do we use?
- ABG and Electrolytes
- What is acidemia vs acidosis?
- What is acidemia?
- pH < 7.35
- What is Alkalemia?
- pH>7.45

Introduction

- What is an anion gap?
- Difference between the serum anions that are measured in a chemistry profile and the unmeasured anions
- What is the normal anion gap?
- 12 +/- 2
- How do you calculate the anion gap?
- Na+ (CI + HCO₃)

The ABG

pH | pCO2 | pO2 | O2 Saturation

- a-A gradiant
- Room Air: 140 (pCO2 + pO2)
- Oxygen:

ACIDEMIA

Respiratory Acidemia

Metabolic

Respiratory Acidosis HCO3 < 24 PCO2 > 40

PCO2 < 40 Alkalosis HCO3 > 24

Anion gap Acidosis "MUDPILES"

- M ethanol
- U remia
- D iabetic Ketoacidosis, Ketoacidosis
- P araldehyde
- I ron, Isoniazid (INH)
- L actic Acidosis
- E thanol, Ethylene glycol
- S alicylates

Signs and Symptoms Gap Acidosis

- Drunk off their
- Hx of drug use
- Fruity breath
- Kussmaul's breathing
- tinnitus
- hypotension

Laboratory Workup

- Chemistries
 - BUN, Cr, glucose
- Lactate level
- Ketones
- Ethanol level
- Salicylate level
- Osmolal gap
- UA

Osmolar Gap Normal (< 25mOsm/kg)

- Uremic Acidosis
- Lactic Acidosis
- Ketoacidosis
- Salicylates

Increased (>25mOsm/kg)

- Ethylene Glycol
 - Look for Oxalate crystals in the Urine

- Methanol Intoxication
 - Visual Changes

Treatment

Treat underlying condition

- Remember:
 - Methanol
 - Ethanol
 - Ethylene Glycol
 - Salicylates
- Can Be Removed via Dialysis

Non gap Acidosis "HARDUPS"

- H yperalimentation
- A cetazolamide, amphotericin
- R TA
- D iarrhea
- U reteral Diversions
- P ancreatic fistula
- S aline resucitation

Is There Intestinal Fluid Loss?

IF YES THINK About

- Ileostomy
- Diarrhea
- Enteric Fistula

Is there Intestinal Fluid Loss?

IF NO: What is the urine pH?

- if > 5.5
 - Type I RTA
- if < 5.5, then CHECK Potassium
- if K is low = RTA type II
- if K is High = RTA type IV

gap? What is the cause?

- P araproteinemias, Multiple myeloma
- L ithium intoxication
- E xcessive Calcium and Magnesium
- A Ibumin is low (hypoalbuminemia)
- B romism

Metabolic Alkalosis Chloride responsive

- Volume Contraction:
 - NG suction
 - Vomitting
 - Diuretics
- Post Hypercapnia
- Hypokalemia
- Hypomagnesemia
- Carbenicillin, Penicillin

Metabolic Alkalosis Chloride Unresponsive

- Adrenal Disorders
 - Glucocorticoid Excess
 - Mineralcorticoid Excess
- Exogenous Steroids
- Alkali Ingestion
- Licorice
- Bartter's Syndrome

Metabolic Alkalosis Signs and Symptoms

- Muscle cramps
- Weakness
- Hypoxia
- Arrhythmias

Metabolic Alkalosis Treatment

- Volume repletion
- Correct Electrolytes
- Spironolactone (hyperaldo)
- Treat Underlying process

Respiratory Acidosis Differential Diagnosis

- Pulmonary Disease
 - Pneumothorax
 - Effusion
 - COPD
 - ARDS
 - PE
 - Inappropriate Vent setting

Respiratory Acidosis Differential Cont'd

- Musculoskeletal Disease
 - Guillain Barre
 - Myasthenia gravis
- CNS
 - Sedatives
 - Trauma
 - Infxn
 - Neoplasm

Respiratory Acidosis Treatment

ADEQUATE VENTILATION

Respiratory Alkalosis Differential Diagnosis

- Pulmonary Disease
 - Pulmonary Edema
 - Pneumonia
 - PE
 - Inappropriate Vent settings
- Why?

Respiratory Alkalosis Differential cont'd

CNS

- Increased Respiratory drive
- Infection
- CVA
- Trauma
- Anxiety
- Drugs
 - Salicylates
 - Catecholamines

Respiratory Alkalosis Differential cont'd

- Sepsis
- Fever
- Pregnancy
- Liver Disease
- Anemia
- Carbon monoxide poisoning

Respiratory Alkalosis Treatment

TREAT UNDERLYING CAUSE

APPROACH TO ACID BASE

Approach to Acid Base

1. Is the patient acid<u>emic</u>(<7.35) or alka<u>lemic</u>(>7.45)

Metabolic	Respirator
HCO3 < 24	PCO2 > 40
HCO3 > 24	PCO2 < 40
	HCO3 < 24

Approach to Acid Base Continued

3. Confirm pH is appropriate for (H+)

$$(H+) = 24 \times pCO2 / HCO3$$

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pH 7.1 7.2 7.3 7.4 7.5 7.6

H+ 78 62 50 40 32 990k to Facts and Formulas
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Acid - Base

4. Calculate the Anion Gap

- Calculate the Anion Gap: (normal is 12 +/- 2)
- If AG is >20 then primary metab gap acidosis is present regardless of pH

Compensated Appropriately

Acid - Rase

```
expected pCO2 = 1.5(HCO3) + 8
met acidosis
                expected pCO2 = .9(HCO3) + 9
met alkalosis
                      each increase in pCO2 of 1,
Acute resp acid
                   pH should decr by .008
Acute resp alk each decrease in pCO2 of 1,
                   pH should incr by .008
                      each increase in pCO2 of 1,
Chronic resp acid
                   pH should decr by .003
Chronic resp alk
                      each decrease in pCO2 of 1,
                   pH should incr by .003
```

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6. The Triple Disorder Calculate the Delta Gap

- If <u>AG</u> acidosis is present: to determine if other underlying condition present
- Take the change in AG= (AG 12)
- Add to HCO3
 - ie (AG 12) + Serum HCO3
- if <23 = non gap acidosis
- if >30 = metabolic alkalosis

EXAMPLES

Examples 1 - 3

- 1. 7.38 | 64 | 146 | 98.8% Na 146, K 3.7, Cl 105, CO2 35
- 2. 7.39 | 65.1 | 62 | 90.6% Na 145, K 4.3, Cl 104, CO2 35
- 3. 7.50 | 34.7 | 58 | 92.4% Na 143, K 3.6, Cl 107, CO2 25

Examples 4 - 6

- 4. 7.32 | 50 | 65 | 95% Na 140, K 3.5, Cl 102, CO2 25
- 5. 7.49 | 34.3 | 121 | 98.6 Na 145, K 3.5, Cl 108, CO2 26
- **6.** 7.41 | 30 | 82 | 96.4 Na 141, K 3.7, Cl 98, CO2 23

Examples 7 - 9

- 7. 7.36 | 46.4 | 175 | 99.1 Na 143, K 4.1, Cl 106, CO2 25
- 8. 7.36 | 50.4 | 166 | 99.0 Na 140, <u>K 2.9</u>, <u>Cl 95</u>, <u>CO2 25</u>
- 9. 7.27 | 44.2 | 159 | 98.8 Na 138, K 3.5, Cl 97, CO2 18

Examples 10 - 12

- 10. 7.495 | 29.8 | 72 | 95.9 Na 137, K 4.1, Cl 95, CO2 22
- **11.** 7.38 | 39.1 | 77 | 95.2 Na 139, K 3.9, Cl 99, CO2 21
- **12.** 7.42 | 38 | 60 | 92 Na 135, K 4.5, Cl 101, CO2 22

Examples 13

13. 7.40 | 30 | 78 | 96.4 Na 147, K 4.0, Cl 15, CO2 12

Diabetic Ketoacidosis

Symptoms of DKA

- Nausea / Vomitting
- Frequent urination
- Thirst -- excessive
- Abdominal pain
- SOB
- Drowsiness
- Stupor
- Unresponsiveness

Signs of DKA

- Hypotension
- Tachycardia
- "fruity" breath
- Kussmaul's respirations
- Orthostasis

Abnormalities of DKA

- Glucose >300
- Serum Bicarbonate <15 mEq/L</p>
- pH <7.30</p>
- Anion gap acidosis
- Electrolyte abnormalities
 - increased potassium

Diabetic Ketoacidosis Treatment

- VOLUME up to 5 6 liters
 - over 2 hours
 - add 5% dextrose when BG < 250 (why?)</p>
- INSULIN
 - 10 units bolus then
 - 5-10 units per hour IV continuous infusion
- CORRECT ELECTROLYTE ABNORMALITIES
 - Hyperkalemia

Diabetic Ketoacidosis Treatment

- BICARBONATE
 - When to administer??

- When do you stop the Insulin Infusion?
- 8-12 hours after anion gap resolved and ketonuria cleared